

THE $\text{Cl}^{35}(\text{n}, \text{p})$ REACTION AND NEUTRON RESONANCE PARAMETERS OF CHLORINE

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The cross section for the $\text{Cl}^{35}(\text{n}, \text{p})$ reaction has been measured as a function of energy for neutrons having energies up to 20 kev. Parameters are given for resonances at -0.21 , 0.405 and 4.3 kev. The results are discussed together with the data previously obtained on the cross section for the radiative capture of neutrons by chlorine.¹

INTRODUCTION

FOR slow neutrons, the (n, p) reaction with chlorine is energetically possible only for the isotope Cl^{35} , whose abundance is 75.4%. For thermal neutrons, the cross section for the reaction $\text{Cl}^{35}(\text{n}, \text{p})$ is about 0.2 barns;²⁻⁵ protons are emitted with an energy of 600 kev. We have measured the cross section for this reaction as a function of energy for neutrons having energies up to 20 kev. In this energy range, the total cross section for the interaction of neutrons with chlorine has been investigated in references 6-8; the cross section for radiative capture has been reported in reference 1.*

METHOD

The measurements were made with a spectrometer based on the slowing down of neutrons in lead.¹⁰⁻¹¹ The energy resolution of the spectrometer (half-width of an isolated resonance) was $\sim 70\%$ for $E = 15$ kev and $\sim 35\%$ for $E \approx 1$ kev. The procedure for measuring a cross section using this method has been described earlier.^{1,9}

Attempts were made to detect the protons with gas discharge counters filled with CCl_4 . The counters were warmed to 50°C (vapor pressure of $\text{CCl}_4 \sim 300$ mm Hg) and operated as proportional counters. Preliminary measurements showed that the reaction $\text{Cl}^{35}(\text{n}, \text{p})$ has resonances at 405 ev and 4.3 kev. However, the poor response time of the counters made quantitative measurements impractical.

The proportional counters were replaced by scintillation detectors. The phosphor was a fine powder of $\text{ZnS}(\text{Ag})$, which was poured into a container filled with CCl_4 or C_3Cl_6 . The latter are transparent liquids which served as both samples

*Preliminary results from reference 1 are quoted in reference 9.

and light pipes.* The effective sample thickness was about 2×10^{21} chlorine nuclei per cm^2 .

The electronic circuitry was set to record protons, but not γ rays, the adjustment being monitored by a strong Co^{60} source ($E_\gamma = 1.3$ Mev). Gamma rays could arise from neutron capture in lead ($E_\gamma = 7.4$ Mev). Other spurious reactions which could have been observed are $\text{Zn}(\text{n}, \alpha)$ and $\text{S}(\text{n}, \alpha)$. The possible contributions of such spurious reactions are indicated by the results shown in Fig. 1.

Even if the extreme assumption were made that the counting rate of the $\text{CCl}_4 + \text{ZnS}$ detector at 30

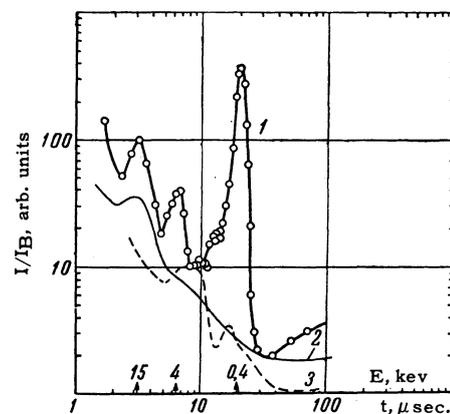


FIG. 1. This figure shows the relative detector counting rate as a function of slowing down time (t). I is the detector counting rate, I_B is the counting rate of a boron counter and is proportional to the neutron density. Curve 1 was taken with a $\text{CCl}_4 + \text{ZnS}(\text{Ag})$ detector, Curve 2 with a $\text{CS}_2 + \text{ZnS}(\text{Ag})$ detector, and curve 3 with a γ ray detector. Curve 3 gives the γ ray background of the spectrometer. Curve 2 is normalized to agree with curve 1 at $t = 30 \mu\text{sec}$, curve 3 is normalized to agree with curve 1 at $t = 9 \mu\text{sec}$.

*Other mixtures were tried, such as $\text{ZnS}(\text{Ag}) + \text{NaCl}$, $\text{CsI}(\text{Tl}) + \text{CCl}_4$, or terphenyl and toluene + NaCl . These were considerably worse from the point of view of efficiency compared to discrimination against γ rays.

μsec were entirely due to the ZnS reaction, while the counting rate at 9 μsec were entirely due to the γ background from radiative neutron capture in lead,* then the contributions of these factors to the counting rate represented by curve 1 at other energies would be negligible.

RESULTS

Figure 2 shows the cross section for the reaction Cl³⁵(n,p), measured as a function of energy. The curve is normalized to the thermal cross section σ_{np} = 0.19 ± 0.05 barn.¹² The uncertainty in the absolute cross section is essentially due to the uncertainty in the thermal cross section, which is ± 26%. Figure 2 also shows the cross section for the (n,γ) reaction in chlorine.¹

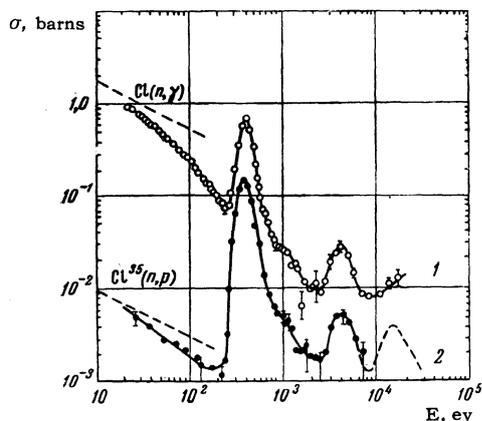


FIG. 2. Energy dependence of the cross sections for neutron capture in chlorine. Curve 1: cross section for the reaction Cl(n,γ), measured with the naturally occurring mixture of isotopes and normalized to a cross section σ_γ = 33.8 barns at E = 0.025 ev.¹² Curve 2: cross section for the reaction Cl³⁵(n,p)S³⁵, normalized so that σ(n,p) = 0.19 ± 0.05 barns at E = 0.025 ev.¹² The errors shown on the curve are statistical; an error also arises because the normalization is uncertain. Taking errors associated with the background also into account, this amounts to ~30%. At energies near 2 kev, the cross section might be a factor 2 less than that shown (because of inadequate corrections for background).

DISCUSSION OF THE RESULTS

1. Negative energy resonance. The dotted straight lines on Fig. 2 represent extrapolations of the thermal (n,γ) and (n,p) cross sections according to the 1/v law. The observed cross section falls off faster than this extrapolation, and

*The γ background due to capture of neutrons in the chlorine sample was small compared to γ-background from neutron capture in lead.

this indicates a resonance at negative energies. This resonance occurs at E₀ = -140 ev according to an analysis of the total cross section⁸ and at E₀ = -210 ev from an analysis of the capture cross section.¹ In the latter analysis, the relation (σ_{γv})^{-1/2} = f(E) was assumed, where according to the Breit-Wigner formula for a narrow resonance (Γ << |E₀|) the function f should be given by

$$(\sigma_{\gamma v})^{-1/2} = c(E - E_0). \tag{1}$$

In the work being reported upon here, the analysis of reference 1 was improved in two ways. In the first place, the experimental values of σ_{γv} were corrected for the presence of the resonance at 405 ev, taking the instrumental resolution into account. The resolution of the apparatus was obtained from measurements of the (n,p) reaction, which showed that the baseline of the peak at 405 ev was very small. In the second place, the capture cross section was corrected for, assuming a 1/v law — i.e., (1) was replaced by the relation

$$[\sigma_{\gamma v} - \alpha(\sigma_{\gamma v_0})]^{-1/2} = c(E - E_0), \tag{2}$$

where (σ_{γv})₀ is the value of σ_{γv} at E = 0.

The experimental data are fitted best by α = 0 and E₀ = -210 ev, although experimental errors are such that α could be less than 4 × 10⁻², which corresponds to |E₀| > 175 ev.* The (n,p) results do not contradict the resonance energy obtained from analysis of the (n,γ) data. However, the value of E₀ obtained from the (n,p) data is unreliable, since it is strongly dependent on the assumptions made about the background due to (n,α) and (n,γ) reactions.

2. Positive energy resonances. Both curves of Fig. 2 clearly show resonances at 405 ev and at 4.3 ± 0.3 kev. There is also a bump near 1 kev which may be due to a resonance at 1.1 ± 0.2 kev. The cross section for the (n,p) reaction showed a peak at ~15 kev; in this energy range, measurements are not reliable because of the large background due to the zinc and sulphur reactions. Since the (n,p) reaction is possible only for Cl³⁵, as was noted above, all the resonances refer to this isotope. In the total cross section measurements, Cl³⁵ resonances have been observed at 405 ev,⁸ 15 kev and 17 kev.¹⁴

Tables I and II show the level parameters calculated from the data on the (n,γ) and (n,p) re-

*Capture by the Cl³⁷ isotope gives α = 0.5 × 10⁻², while resonances with positive energy give α < 0.1 × 10⁻². The direct capture of neutrons must follow the 1/v law,¹³ but there are no data on the magnitude of direct capture for chlorine.

Table I. Parameters of neutron resonances in Cl³⁵

Parameter	E ₀ , ev				
	-210±10	405	1100±200	4300±300	(15 and 17)×10 ³
Γ _γ , ev	0.50±0.01	0.5*	0.5*	0.5*	
σ ₀ Γ _γ , barn-ev		120±10	~8	80±27	
σ ₀ Γ _p , barn-ev		16±5	~0.8	5.6±2.0	~10
10 ³ Γ _p , ev	2.4±0.8	70±22	~50	35±15	~100
γ _{0p} ² , ev**	70	[2.10 ³]	[~10 ³]	[8.10 ²]	~2.10 ³
γ _{1p} ² , ev**		9.10 ³	~6.10 ³	3.8.10 ³	
10 ³ Γ _n , ev***		26-65	4-30	250-700	(30+35)·10 ³ [14]
γ _{0n} ² , ev**	7.4·10 ²	[1.0-1.7]	[0,1-0,2]	[3.3-5.8]	(1.2+1.4)·10 ²
γ _{1n} ² , ev**		(2-4.9)·10 ³	(1-5)·10 ²	(5.6-16)·10 ²	
l	0 (g=5/8)	1	1	1	0

*The radiation widths are taken to be equal to the radiation width of the level at E₀ = -210 ev.
 **γ_{0p}² and γ_{0n}² are the reduced proton and neutron widths calculated on the assumption that the resonance corresponds to orbital angular momentum l = 0; similarly, γ_{1p}² and γ_{1n}² correspond to the assumption l = 1.
 ***The magnitude of Γ_n depends on the assumption made about the spin of the compound nucleus Cl³⁶ (see Table II).

Table II. Parameters of the resonances at 405 ev and 4.3 kev in Cl³⁵ as a function of the statistical factor g = (2J + 1)/2 (2l + 1)

E ₀	g*	10 ³ Γ _n , ev	10 ³ gΓ _n , ev	Γ, ev	σ ₀ Γ, barn-ev	σ, бн
405 ev	3/8	65±5	25±2	0,64	155±15	240±25
	5/8	38±3	24±2	0,61	147±15	240±25
	7/8	26±2	22±2	0,60	144±15	240±25
4,3 kev	3/8	> 700	> 260	> 1,2	> 160	~140
	5/8	410±150	260±80	0,95	155±40	164±40
	7/8	250± 50	220±50	0,78	130±27	167±40

*The value g = 1/8 for the 405 ev level is inconsistent with the measured cross section for absorption, while for the 4.3 kev level it does not accord with measurements of the (n, γ) cross section.

actions. In the calculations the radiation widths of the various levels were taken equal to the radiation width of the level at -210 ev, i.e., equal to 0.5 ev.¹ Supplementary measurements of self-absorption were carried out for the resonance at 405 ev. These gave a value for the strength σ₀Γ of the level of 120 ± 30 barn-ev per Cl³⁵ nucleus, which agrees with the data of reference 1 (see Table II).

3. The negative energy resonance in Cl³⁵ determines the cross section for capture of thermal neutrons and hence must be an s-resonance, i.e., corresponds to capture of a neutron with orbital angular momentum l = 0. If the resonances at 405 ev, 1.1 kev and 4.3 kev were also taken to correspond to l = 0, then the reduced neutron widths of these resonances* would come out to be

*The reduced partial widths γ₁₁² were calculated by dividing the measured partial widths Γ₁₁ by twice the penetrability P_l: γ₁₁² = Γ₁₁/2P_l, where P_l = kR/F_l² + G_l² (for further details, see reference 15). The calculations were carried out using the tables in reference 16 and the formula R = 1.3 A^{1/2} × 10⁻¹³ cm.

two to three orders of magnitude smaller than the reduced neutron widths for the resonances at -210 ev and 17 kev (the last is an s-resonance¹⁴). Hence the resonances at 405 ev, 1.1 kev and 4.3 kev must have l = 1. For the 405 ev resonance, this conclusion contradicts the data quoted in reference 14; the 1.1 kev and 4.3 kev resonances are not noted in reference 14.

It should be noted that the reduced proton width for the negative energy resonance is very small.

In the energy range 0.5 to 2 × 10⁴ ev the reso-

nance integral $R = \int_{E_1}^{E_2} \sigma dE/E$ for the Cl³⁵ (n, p)

reaction is equal to R_p = 0.074 ± 0.025. For the (n, γ) reaction this integral is R_γ = 14.0 ± 0.8 b.¹ The total resonance integral for absorption is thus equal to 14.1 ± 0.8 b, which is in satisfactory agreement with results obtained by other authors (12 barns¹⁷ and 12.7 ± 1.7 barns¹⁸).

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¹Kashukeev, Popov, and Shapiro, *Нейтронная физика (Coll., Neutron Physics) Atomizdat*, in press; *J. Nucl. Energy* (to be published).

²Gilbert, Roggen, and Rossel, *Helv. Phys. Acta* **17**, 97 (1944).

³M. Maurer, *Z. Naturforsch.* **4**, 150 (1949).

⁴Seren, Friedlander, and Turkel, *Phys. Rev.* **72**, 888 (1947).

⁵H. Berthet and J. Rossel, *Helv. Phys. Acta* **27**, 159 (1954), *Helv. Phys. Acta* **28**, 265 (1955).

⁶C. T. Hibdon and C. O. Muehlhause, *Phys. Rev.* **79**, 44 (1950).

⁷Toller, Patterson, and Newson, *Phys. Rev.* **99**, 620 (1955).

⁸Brugger, Evans, Joki, and Shankland, *Phys. Rev.* **104**, 1054 (1957).

⁹Bergman, Isakov, Popov, and Shapiro, *Ядерные реакции при низких и средних энергиях (Nuclear Reactions at Low and Medium Energies)*, Academy of Sciences, U.S.S.R., 1958 (p. 141).

¹⁰Lazareva, Feinberg, and Shapiro, *JETP* **29**, 381 (1955), *Soviet Phys. JETP* **2**, 351 (1956).

¹¹Bergman, Isakov, Murin, Shapiro, Shtranikh, and Kazarnovskii, *Proc. Geneva Conference* **4**, 166 (1955).

¹²D. J. Hughes and R. B. Schwartz, *Neutron Cross Sections*, BNL-325 (second edition, 1958).

¹³A. M. Lane and J. E. Lynn, *Nucl. Phys.* **17**, 563 (1960).

¹⁴Hughes, Magurno, and Brussel, *Neutron Cross Sections*, BNL-325 (Suppl. 1 to second edition, 1960).

¹⁵A. M. Lane and R. G. Thomas, *Rev. Modern Phys.* **30**, 257 (1958).

¹⁶Bloch, Hull, Broyles, Bouricius, Freeman, and Breit, *Revs. Modern Phys.* **23**, 147 (1951).

¹⁷P. Macklin and H. Pomerance, *Proc. Geneva Conference* **5**, 96 (1955).

¹⁸V. B. Klimentov and V. M. Gryazev, *Атомная энергия (Atomic Energy)* **3**, 507 (1957).

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